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chemistry. • Describe oxidation and reduction (redox) reactions. • Discuss the different forms of matter: elements, compounds, and mixtures. • Explain the difference between solutions, suspensions, and emulsions. • Explain pH and the pH scale.

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**Organic chemistry.** The study of substances that contain the element carbon. **oxidation.** A chemical reaction that combines a substance with oxygen to produce an oxide. **oxidation-reduction.** Also known as redox; a chemical reaction in which the oxidizing agent is reduced and the reducing agent is oxidized. **oxidizing agent.**

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STUDY. PLAY. Chemistry. the science that deals with the composition, structures, and properties of matter and how matter changes under different conditions. **Organic Chemistry.** the study of substances that contain the element carbon. **Inorganic chemistry.**

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PLAY. acidic solution. A solution that  
has a pH below 7.0 (neutral). alkaline  
solution. A solution that has a pH  
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A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. (aka redox) a chemical reaction in which the oxidizing agent is reduced (by losing oxygen) and the reducing agent is oxidized (by gaining oxygen). oxidizing agent. Substance that releases oxygen.

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A stable physical mixture of 2 or more substances. ALKANOLAMINES. Alkaline substances used to neutralize acids or raise the PH of many hair products. EMULSION. An unstable physical mixture of 2 or more immiscible substances, plus a special

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Chemistry ingredient called a "emulsifier".

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