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acids A. bitter L2. pH B. ionize only
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H₂ (g) F 4. neutralization D. salt +...

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(blank) taste. Sour Taste. What color
does litmus paper turn when it's
dipped in an acid. Red. What color
does litmus paper turn when it's
dipped in a base.

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Baylor, Scott / Chapter 15 Acids and Bases Chapter 15 Test Acids And Bases Answers when acids or bases act as acids or bases, they form their conjugate and the strength is the inverse of their parent. and there are certain strong acids and strong bases.

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LiOH, NaOH, KOH - when act as bases (get rid of H^+ ion) then left with
Li Page 13/24

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Define acid and base according to Arrhenius's theory of ionization. Explain the differences between strong and weak acids and bases. FIGURE 15-1 (a) Fruits and fruit juices contain acids such as citric acid and ascorbic acid. Carbonated beverages contain benzoic acid, phosphoric acid, and

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Carbonic acid. (b) Many household cleaners contain ...

CHAPTER 15 Acids and Bases

Chapter 15: Acids and Bases
Acids and Bases Arrhenius Definitions: acids
- compounds that produce an increase
in $[H^+]$ when dissolved in water bases
- compounds that produce an increase
in $[OH^-]$ when dissolved in water Lewis
Definitions: acids - electron pair
acceptors bases - electron pair donors
Brønsted-Lowry Definitions: acids - H^+
donors

Chapter 15: Acids and Bases Acids and Bases

a. Acid Base Practice Test Answer
Section MULTIPLE CHOICE 1. ANS:
B DIF: I OBJ: 15-1.1 2. ANS: C DIF: I
OBJ: 15-1.1 3. ANS: A DIF: I OBJ:
15-1.1

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Quiz 1; Practice Quiz 2; Practice Quiz

3. Chapter 16: Acid-Base Equilibria

Practice Quiz 1; Practice Quiz 2;

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& Complex Ion Equilibria ...

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a. a strong acid b. a weak acid c. a strong base d. a weak base e. neither an acid nor a base 24) A 25.0-mL sample of a solution of a monoprotic acid is titrated with a 0.115 M NaOH solution. The titration curve above was obtained. The concentration of the monoprotic acid is about _____ mol/L.

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