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select Copper.

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Answer: A.

Substance A A

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Chemist mixes

500 g of lead at
500°C with 1,200

g of water at
20°C. She then

mixes 500 g of
copper at 500°C
with 1,200 g of
water at 20°C.

The specific
heat capacity of
lead is 0.1276
J/g°C and the
specific heat

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capacity of
copper is 0.3845
 $\text{J/g}^\circ\text{C}$.

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of the Answer

substances can
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specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can

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be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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the lead-water
system will be
lower than the
final
temperature of

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